# MASSACHUSETTS INSTITUTE OF TECHNOLOGY <br> Department of Civil and Environmental Engineering <br> 1.77 Water Quality Control <br> Spring 2006 

Final Exam
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Open Book

## Problem 1

An oil spill occurs 30 km off the coast of California where the prevailing winds are from the west at $5 \mathrm{~m} / \mathrm{s}$ and the ambient current flows to the south at a speed of 10 $\mathrm{cm} / \mathrm{s}$. Approximate the trajectory of the spill during the first 24 hours; i.e. how far will it have traveled and in what direction?

## Problem 2

A tropical lake has the idealized two-layer geometry shown in the figure. The top layer has horizontal area of $2 \times 10^{6} \mathrm{~m}^{2}$ and thickness 4 m , while the bottom layer has horizontal area of $10^{6} \mathrm{~m}^{2}$ and thickness 4 m . If the temperature of the top layer is $23^{\circ} \mathrm{C}$ and the temperature of the bottom layer is $20^{\circ} \mathrm{C}$, compute the lake stability index. Assume the following equation of state for freshwater density ( $\rho$ ) in $\mathrm{g} / \mathrm{cm}^{3}$ as a function of temperature T in ${ }^{\circ} \mathrm{C}$ :

$$
\rho=1-0.0000063(T-4)^{2}
$$

Starting with the temperatures described above, the lake experiences steady meteorological conditions characterized by an equilibrium temperature $\mathrm{T}_{\mathrm{e}}$ of $10^{\circ} \mathrm{C}$ and a surface heat exchange coefficient K of $30 \mathrm{~W} / \mathrm{m}^{2}-{ }^{\circ} \mathrm{C}$. Neglecting internal mixing, how long would it take for the lake to become isothermal? Note: the specific heat of water $\mathrm{c}_{\mathrm{p}}$ $=4182 \mathrm{~J} / \mathrm{kg}-{ }^{\circ} \mathrm{C}$.


## Problem 3

A small tidal embayment receives a steady freshwater discharge of $0.01 \mathrm{~m}^{3} / \mathrm{s}$ that contains suspended solids with concentration of $20 \mathrm{mg} / \mathrm{L}$. The embayment has a surface area of $1 \mathrm{~km}^{2}$ and an average depth of 2 meters. During a period of one week, dye was added continuously to the inflow at a rate of $0.01 \mathrm{~g} / \mathrm{sec}$. At the end of the week, measurements within the embayment revealed an average dye concentration of $10^{-3} \mathrm{mg} / \mathrm{L}$ and an average particle concentration of $10^{-2} \mathrm{mg} / \mathrm{L}$. (The measurements were averaged spatially over the embayment and temporally over one tidal cycle.) Based on this information compute the hydrodynamic residence time of the dye and the average fall velocity of the particles. The inflow contains trace amounts of a chemical which is sorbed to the particles. Assuming equilibrium partitioning with a solid-water partition coefficient of $10^{4} \mathrm{~cm}^{3} / \mathrm{g}$, estimate the percentage of the chemical in the sorbed phase in the inflow and in the embayment.

