SCATTERING

INTERACTION OF RADIATION WITH MATTER

CHARGED PARTICLES IN MATTER

□ Charged particles interact mostly with the electronic cloud



Small energy loss, but very frequent collisions

CHARGED PARTICLES IN MATTER

- Classical, non-relativistic collisions of charged particles with electrons
- □ Conservation of energy and momentum



□ Charged particle looses a tiny fraction of its original energy

$$\Delta E = \frac{1}{2}m_e v_e^2 = \frac{1}{2}m_e (2v_\alpha)^2 = 4\frac{m_e}{m_\alpha}E_\alpha \quad \rightarrow \quad \frac{\Delta E}{E_\alpha} = 4\frac{m_e}{m_\alpha} \ll 1$$

COULOMB INTERACTION



STOPPING POWER

□ Integrate over all impact parameters b

$$-dE = 2\pi dx \int n_e \Delta Ebdb \quad \rightarrow \quad \frac{dE}{dx} = -2\pi \int n_e \Delta E(b)bdb = -\frac{4\pi e^4 Z_\alpha^2 n_e}{(4\pi\epsilon_0)^2 m_e v_\alpha^2} \int \frac{db}{b}$$

□ Lower bound (closest approach max energy lost):

$$b_{min} \sim \frac{1}{4\pi\epsilon_0} \frac{e^2}{2m_e v_\alpha^2}$$

□ Upper bound: Bohr radius (from ionization energy)

$$\frac{b_{max}}{b_{min}} = \frac{2m_e v_\alpha^2}{Z_\alpha E_I}$$

$$-\frac{dE}{dx} = \frac{4\pi e^4 Z_\alpha^2 n_e}{(4\pi\epsilon_0)^2 m_e v_\alpha^2} \ln\left(\frac{b_{max}}{b_{min}}\right) = \frac{4\pi e^4 Z_\alpha^2 n_e}{(4\pi\epsilon_0)^2 m_e v_\alpha^2} \ln\Lambda$$

STOPPING RANGE

- Thousands of events (collisions) are needed to effectively slow down and stop the alpha particle
- 2. As the alpha particle is barely perturbed by individual collisions, the particle travels in a straight line.
- 3. The collisions are due to Coulomb interaction, which is an infinite-range interaction. Then, the alpha particle interacts simultaneously with many electrons, yielding a continuous slowing down a certain stopping range.
- 4. The electrons which are the collision targets get ionized, thus they lead to a visible trail (e.g. in cloud chambers)

RUTHERFORD SCATTERING ANIMATION



RUTHERFORD SCATTERING











Why Rutherford used Gold in the experiment?

MIT OpenCourseWare http://ocw.mit.edu

22.02 Introduction to Applied Nuclear Physics Spring 2012

For information about citing these materials or our Terms of Use, visit: http://ocw.mit.edu/terms.